

Safety Data Sheet Revision date: 12/1/2022

1. Identification

Product identifier 12.5 – 15.6% Solution

Other means of identification

Synonyms Bleach 12.5% - Sodium Hypochlorite Solution, Bleach Solution, Liquid Chlorine Solution,

Hypo-solution and Liquid Bleach.

Manufacturer/Importer/Supplier/Distributor Information

Company name INEOS KOH
Address 3509 Middle Road
Ashtabula, OH 44004

Telephone

Emergency phone number 1-800-424-9300 (24 hours)

CHEMTREC

1-440-997-5221

2. Hazard(s) identification

Label elements



Signal word DANGER

Hazard statement Causes severe skin burns and eye damage.

Very toxic to aquatic life.

Precautionary statement

Prevention Do not breathe mist, vapors, or spray.

Wash hands thoroughly after handling. Avoid release to the environment.

Wear protective gloves/protective clothing/eye protection/face protection.

Response If Swallowed: Rinse mouth. DO NOT induce vomiting.

IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with

water/shower.

Wash contaminated clothing before re-use.

IF INHALED: Remove person to fresh air and keep comfortable for breathing. Immediately call a POISON CONTROL CENTER or doctor/physician. Specific treatment (see First Aid Measures on Safety Data Sheet).

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present

and easy to do. Continue rinsing

Collect spillage. Store locked up

Disposal Dispose of contents/containers in accordance with local/regional/national/international regulations.

Hazard(s) not otherwise classified (HNOC)

Supplemental Information Reacts violently with acids liberating chlorine gas. Also reacts with organic substance. When

heated, gives off oxygen that may increase fire hazard

NFPA classification (scale 0-4):

Storage

Health 2
Fire 0
Reactivity 1

EC classification (assigned): C (Corrosive)

Emergency overview

Major health hazards: Respiratory Tract Burns, Skin Burns, Mucous Membrane Burns, and Eye Irritation

Potential health effects

SODIUM HYPOCHLORITE Revision Date: 12/01/2022

Irritation to respiratory tract. May have same as effects reported in other routes of exposure, burns, Short term exposure

blisters, nausea, difficulty breathing, and lung congestion.

Long term exposure

Skin contact:

Irritant, reddening of the skin. May have burns, blisters, and itching

Short term exposure

Long term exposure

Same as effects reported in short term exposure.

Same as effects reported in short term exposure.

Eve contact:

Short term exposure Long term exposure

Irritation (possible severe), possible eye damage Same as effects reported in short term exposure.

Ingestion:

Short term exposure Long term exposure

Burns, vomiting stomach pain, disorientation, bluish skin color, convulsions, and coma

Same as effects reported in short term exposure.

Carcinogen status

OSHA: No NTP: No IARC: No

3. Composition/information on ingredients

Mixtures

Chemical name	CAS number	%
Sodium Hypochlorite (NaOCl)	7681-52-9	12.5 - 15.6
Sodium Hydroxide (NaOH)	1310-73-2	0.2 - 1.5
Water (H ₂ O)	7732-18-5	Balance

4. First-aid measures

Inhalation Remove from exposure and get fresh air. Use bag valve mask or similar device to perform

artificial respiration (rescue breathing) if needed. Keep warm and at rest. Get medical attention

immediately if artificial respiration required

Skin contact Remove contaminated clothing, jewelry, and shoes immediately. Flush affected area with large

> amounts of water, preferable a safety shower. Use soap or mild detergent and large amounts of water until no evidence of chemical remains (at least 15-20 minutes). For burns, cover affected area securely with sterile, dry, loose fitting dressing. If skin is burned, get medical attention

immediately.

Eye contact Wash eyes immediately with large amounts of water, occasionally lifting upper and lower lids,

> until no evidence of chemical remains (at least 15 minutes). Continue irrigating with a normal saline solution until ready to transport to physician. Cover with sterile bandages. Get medical

attention immediately

Ingestion Rinse mouth with water. Drink large quantities of milk (water if no milk is available). Milk of

magnesia may be helpful. DO NOT USE ACIDIC ANTIDOTES SUCH AS SODIUM BICARBONATE. When vomiting occurs, keep head lower than hips to help prevent aspiration. If person is unconscious, do not induce vomiting and turn their head to the side. Never make an

unconscious person vomit or drink fluids. Get medical attention.

General information NOTE TO PHYSICIAN: For inhalation, consider oxygen. For ingestion, avoid gastric lavage,

emesis, sodium bicarbonate and acidic solutions. Consider the use of antacids.

5. Fire-fighting measures

Suitable extinguishing media

Specific hazards arising from

the chemical

Fire-fighting

equipment/instructions

Regular dry chemical, carbon dioxide, water, or foam suitable for surrounding fire. For large fires, use regular foam or flood with fine water spray.

Negligible fire hazard. Oxidizer, this material will react with some metals and cause liberation of oxygen. May ignite or explode on contact with combustible materials. Toxic fumes can be liberated by contact with acid or heat.

Wear self-contained breathing apparatus and full protective clothing. Move container from fire area if it can be done without risk. Cool containers with water spray until well after the fire is out. Stay away from the ends of tanks. Use extinguishing agents appropriate for surrounding fire. Do not get water directly on material. For large fires, flood with fine water spray. Reduce vapors with water spray. Apply water from a protected location or from a same distance. Avoid body contact

or inhalation of material or combustion by -products. Stay upwind and keep out of low areas.

SODIUM HYPOCHLORITE Revision Date: 12/01/2022

6. Accidental release measures

Personal precautions, protective equipment and emergency procedures Methods and materials for

Do not touch spilled material. Stop leak if possible without personal risk, keep unnecessary people away, isolate hazard area and deny entry.

For small spills, collect spilled material in appropriate container for disposal and consider containment and cleaning up absorbing with sand or other non-combustible material (e.g., do not use sawdust or other combustible material). Be advised, however, that the use of absorbing material is creating hazardous waste and this absorbing material must now be disposed of properly. Collect spilled material in appropriate container for disposal. For small dry spills, move containers away from spill to a safe area. For large spills, dike for later disposal. Contain in as small an area as possible, such as a holding area for dilution and neutralization. Contain spill in plastic drums when available. Dispose of in accordance with Federal, State, and local regulations. Personnel engaged in cleanup operations must be equipped with NIOSH approved respirator protection, rubber boots, gloves, and clothing to avoid body contact. Reportable Quantity (RQ): 100 pounds. Notify Local Emergency Planning Committee and State Emergency Response Commission for release greater than or equal to RQ (U.S. SARA Section 394). If release occurs in the U.S. and reportable under CERCLA Section 103, notify the National Response Center at (800)424-8802 (USA) or (202)426-2675 (USA).

Environmental precautions

If possible, do not allow material to enter sewers, streams, ponds or storm conduits, as

concentrated solutions will seriously injure aquatic life.

Advance planning

Plan in advance for an occupational release and have necessary equipment and neutralization

agents on-site. Contact Odyssey Manufacturing for assistance.

Dot emergency guide number

7. Handling and storage

Precautions for safe handling Store in vented, closed containers that provide protection from direct sunlight. Store and handle in

accordance with all current regulations and standards including NFPA 430 Code for the Storage of Liquid and Oxidizing materials. Do not mix old bleach with new bleach.

Condition for safe storage, including any incompatibilities

Keep separated from incompatible substances and do not store near acids, heat, or oxidizable materials or organics. When handling, do not mix with other cleaning agents that may liberate chlorine gas vapors (e.g., acidic agents). Contact with metals will accelerate decomposition

and generate oxygen potentially generating pressure in closed vessels.

8. Exposure controls/personal protection

154

Occupational exposure limits

2 mg/m³ AIHA recommended STEL 15 minutes for Sodium Hypochlorite

Appropriate engineering controls Provide local exhaust ventilation system. Ensure compliance with applicable exposure limits. Individual protection measures, such as personal protective equipment

Eye/face protection

Splash goggles are preferred to a face shield. Another option is to wear splash resistant safety goggles with a face shield. Provide an emergency eye wash fountain and quick drench shower in the immediate work area.

Skin protection Hand protection Other

Wear appropriate chemical resistant gloves.

It is recommended to wear appropriate chemical resistant clothing to avoid body contact such as a rubber apron or rain suit. Boots are preferred footwear.

Respiratory protection

Any self-contained breathing apparatus with a full face piece (use for High Concentrations or those which are Immediately Dangerous to life or Health). Under conditions of frequent use or heavy exposure, respiratory protection may be needed. Respiratory protection is ranked in order from minimum to maximum. Consider warning properties before use.

- Any chemical cartridge respirator with organic vapor cartridge(s).
- Any chemical cartridge respirator with a full facepiece and organic vapor cartridge(s).
- Any air-purifying respirator with a full facepiece and an organic vapor canister.
- Any supplied-air respirator with a full facepiece and operated in a pressure-demand or other positive pressure mode in combination with a separate escape supply (use for Unkonwn Concentrations of those that may be Immediately Dangerous to Life or Health)

9. Physical and chemical properties

Appearance

Physical state Liquid
Form Liquid
Color Pale yellow

Odor Chlorine odor like household bleach

Odor threshold 0.9 mg/m³ **pH** 11.5 – 13 (68⁰F)

Melting point/freezing point - 10⁰F
Initial boiling point and Not available

boiling range

Flash point Not applicable
Evaporation rate No data available
Flammability (solid, gas) Not available
Upper/lower flammability or explosive limits
Flammability limit – Not available

lower (%)

Flammability limit – Not available

lower (%) temperature

Flammability limit – Not available

upper (%)

Flammability limit – Not available

upper (%) temperature

Explosive limit – Not available

lower (%)

Explosive limit – Not available

upper (%)

Vapor pressure Vapor Pressure of water + decomposition product vapor pressure

Vapor densityNot AvailableRelative density $1.19 - 1.28 (60^{0}F)$

Solubility(ies)

Solubility (water) Completely miscible

Partition coefficient Not available

(n-octanol/water)

Auto-ignition temperature Not applicable

Decomposition temperature Not available (The degradation rate doubles for every 10^{0} F above 70^{0} F)

Viscosity Not Available

Other information

Molecular weight 74.44 Chemical family Alkali

10. Stability and reactivity

Reactivity Stable at normal temperature and pressure. Will slowly decompose over time. Rate of decomposition is

dependent upon initial strength, temperature, metal contaminants and sunlight exposure.

Conditions to avoid Avoid heat, flames, sparks and other sources of ignition. Dangerous gases may accumulate in confined spaces. May ignite or explode on contact with combustible materials. Contamination with metals will

decompose to form sodium chloride and oxygen gas.

Incompatible materials Acids, metals, amines, combustible materials, reducing agents. Specific reactions with sodium

Hypochlorite include the following: ACIDS: Violent reaction, ALUMINUM: Corrosive action, AMINES: Form explosive chloramines, AMMONIA: Form explosive chloramines, AMMONIUM SALTS: May form explosive product, BENZYL CYANIDE (ACIDIFIED): explosive reaction, ETHYLENEAMINE: Forms explosive 1-chloroethyleneamine, FORMIC ACID: Explosive mixture, METHANOL: May form explosive compound, NITROGEN COMPOUNDS: Forms explosive N-chloro compounds, ORGANIC AND COMBUSTIBLE MATERIALS: Fire and explosion hazard, OXALIC ACID: Intense reaction, REDUCING AGENTS: Fire and explosion

hazard, ZINC: Corrosive

Hazardous decomposition Thermal decomposition products-Chlorine and Hydrochloric Acid Vapors.

products Decomposition products - Hypochlorous Acid Vapors

Polymerization Will not polymerize

SODIUM HYPOCHLORITE Revision Date: 8/3/2022

Page 4 of 7

11. Toxicological information

Information on likely routs of exposure

Ingestion

Acute May cause irritation and erosion of the mucous membranes, vomiting (possibly bloody) and

abdominal pain and spasms. A drop in blood pressure, shallow respiration, edema (possibly severe) of pharynx, larynx, and glottis, confusion, convulsions, delirium and coma may occur. Cyanosis and circulatory collapse are possible. Esophageal or gastric perforation and strictures are rare. Death may occur, usually due to complications of severe local injury such as toxemia, shock, perforations, hemorrhage, infection and obstruction. Massive ingestions may produce fatal

hyperchloremic metabolic acidosis or aspiration pneumonitis.

Chronic Sensitization reactions are reported in individuals who are exposed in small amounts through their

water supply. High doses have caused sperm abnormality in mice

Inhalation

Acute May cause severe bronchial irritation, sore throat with possible blistering, coughing, stomatitis,

nausea labored breathing, shortness of breath and pulmonary edema. 10-20 mg/m³ causes

burning of the nose and throat; 40-60 mg/m³ may be fatal. If sufficient amounts are absorbed, may

cause effects as detailed in acute ingestion.

Chronic No data available

Skin contact

Acute Extent of damage depends on concentration, pH, volume of solution, and time of contact. May

cause redness, pain, blistering, itchy eczema & chemical burns. Sensitization reactions possible in

previously exposed persons.

Chronic Allergic dermatitis has also been reported

Eye contact

Acute May cause redness, pain, & blurred vision. Solutions of 5% splashed in human eyes have caused a

burning sensation and later only slight superficial disturbance of the corneal epithelium, which cleared completely in the next day or two without special treatment. However, one animal study reports a 5% solution causing only moderate irritation with clearing within 7 days. A higher concentration of 15% tested on rabbit eyes caused immediate severe pain, hemorrhages, rapid onset of ground-glass appearance of the corneal epithelium, moderate bluish edema of the whole cornea, chemosis and discharge for several days. Such eyes have sometimes healed in 2-3 weeks with slight or no residual corneal damage but they had neovascularization of the conjunctiva and

distortion of the nictitating membrane by scarring.

Chronic Depending on concentration and time of exposure, symptoms may be as those of acute exposure.

Information on toxicological effects

Acute Toxicity

TestSpeciesTest resultsOral-TDLoWoman1gm/kgOral-LD50Mouse5800 mg/kgIntravenous-TDLoMan45 mg/kgOral continuous-TDLoRat140 mg/kg/9 weeks

Carcinogenicity According to the IARC, animal inadequate evidence, human no adequate data, Group 3

(Hypochlorite salts)

Irritation data 10 mg eyes-rabbit moderate

Local effects Corrosive: inhalation, skin contact, eye, ingestion hazards

Acute toxicity level Slightly Toxic if ingested

Mutagenic data Mutation in micro organisms-Salmonella typhimurium 1 mg/plate (-S9); DNA repair-

Escherichiacoli 20 μg/disc; DNA damage-Escherichiacoli 420 μmol/L: phage inhibition capacity-Escherichiacoli 103 μg/well; micronucleus test-non=mammalian species multiple 200 ppb; cytogenetic analysis-non-mammalian species multiple 120 μg/L; cytogenetic analysis-hamster

 $lung\ 100\ mg/l$

12. Ecological information

Ecotoxicity

Fish toxicity LC50 (Mortality) Cutthroat trout (Oncorhynchus clarki) 94.0 µg/L 96 hour(s) Invertebrate toxicity LC50 (Species Diversity) protozoan phylum (Protozoa) $31.6 \mu g/L 7 hour(s)$ 0.37 - 2.3 ppm (48 hour) Daphnia magna LC50 Daphnia magna Algal toxicity LC50 (Mortality) Algae, phytoplankton, algal mat (Algae) 90 μg/L 96 hour(s) Phyotoxicity Curled pondweed (Potamogeton crispus) 230μg/L 35 hour(s) (Biomass) Other toxicity (Chlorophyll) Aquatic community (Aquatic community) $2.1 \,\mu g/L \, 28 \, day(s)$

SODIUM HYPOCHLORITE Revision Date: 8/3/2022

Page 5 of 7

13. Disposal considerations

Disposal instructions Dispose in accordance with all applicable regulations. Subject to disposal regulations: U.S. EPA

40 CFR 262.

Hazardous Waste Number(s): D001.

14. Transport information

DOT

UN number UN1791

UN proper shipping name Sodium Hypochlorite

Transport hazard class(es)

Class 8

Subsidiary risk

Packing group III (less than 16% available chlorine)/ II (16% or more available chlorine)

Marine pollutant Yes

Packaging exceptions 49 CFR 173.154

Packaging non bulk 49 CFR 173.203 (less that 16% available chlorine)/ 49 CFR 173.202 (16% or more available

chlorine)

Packaging bulk 49 CFR 173.241 (less that 16% available chlorine)/ 49 CFR 173.242 (16% or more available

chlorine)

U.S. DOT 49 CFR 172.101 and subpart E

Labeling requirements Corrosive

U.S. DOT 49 CFR 172.101 quantity limitations

Passenger aircraft or railcar 5 LITERS (less that 16% available chlorine) / 1 LITERS (16% or more available chlorine) **Cargo aircraft only**: 60 liters (less that 16% available chlorine) / 30 liters (16% or more available chlorine)

15. Regulatory information

U.S. regulations

TSCA inventory status Yes **TSCA 12(b) export** Not listed.

notification

CERCLA section 103 Yes

(40CFR302.4)

Sodium hypochlorite 100 lbs RQ

Superfund Amendments and Reauthorization Act of 1986 (SARA)

SARA section 302 No

(40CFR355.30)

SARA section 304 No

(40CFR355.40)

SARA section 313 No

(40CFR372.65)

SARA hazard categories, SARA sections 311/312 (40CFR370.21):

Acute Yes
Chronic No
Fire No
Reactive No
Sudden release No
OSHA process safety No

(29CFR1S1O.19)

State regulations
California Proposition 65 No

European regulations

EC Number (BINECS) 231-668-3

EC risk and safety phrases:

R 31 Contact with acids liberates toxic gas.

R 34 Causes burns

 $S^{1/2}$ Keep locked-up and out of reach of children.

S 28b After contact with skin, wash immediately with plenty of soap and water.

S 45 In case of accident or if you feel unwell, seek medical advice immediately (show the label where

possible)

SODIUM HYPOCHLORITE

Revision Date: 8/3/22 Page 6 of 7

S 50 Do not mix with incompatible materials.

Concentration limits:

C>10% C R 31-34 5%<=C<=10% Xi R 31-36/38

German regulations

Water hazard class (WGK) 2 (Official German Classification)

FIFRA information INEOS KOH product is not registered as a pesticide with the EPA under FIFRA

regulations. This chemical is subject to certain labeling requirements under federal pesticide law. These requirements differ from the classification criteria and hazard information required for safety data sheets, and for workplace labels of non-pesticide chemicals. Following is the

hazard information as required on the pesticide label:

Warning May be fatal if swallowed or if absorbed through skin.

Harmful if inhaled. Causes skin irritation.

Causes substantial but temporary eye injury. This pesticide is extremely toxic to fish

16. Other information

For additional information, contact our Quality Assurance / Technical Service Department.

Information contained in this MSDS refers only to the specific material designated and does not relate to any process or use involving other materials. This information is based on data believed to be reliable, and the Product is intended to be used in a manner that is customary and reasonably foreseeable. Since actual use and handling are beyond our control, no warranty, expressed or implied, is made and no liability is assumed by INEOS KOH in connection with the use of this information.

SODIUM HYPOCHLORITE Revision

Date: 8/3/2022