

#### **SODIUM SULFITE SOLUTION**

# **Section 1 - Product and Company Identification**

Product Name: Sodium Sulfite Solution

**Chemical Formula:** Na<sub>2</sub>SO<sub>3</sub> CAS Number: 00777-83-7

General Use: Waste water dechlorination agent, lab reagent, reducing agent

Manufacturer: INEOS Calabrian Corporation

5500 Hwy. 366

Port Neches, Texas77651

**Telephone:** 409-727-1471 **Fax:** 409-727-5803

Emergency Contact: CHEMTREC 800-424-9300

## Section 2 - Hazards Identification

**Emergency Overview** 

Target Organs: Eyes, skin and respiratory

**GHS Classification:** Eye – Causes eye irritation (Category 2B)

Skin - Causes skin irritation (Category 2)

Inhalation – May be harmful if inhaled (Category 5)

GHS Label Elements: Signal Word - Warning

Pictogram



Irritant

Hazard Statements: H315 – Causes skin irritation

H320 – Causes eye irritation. H333 - May be harmful if inhaled.

**Precautionary** P264 – Wash thoroughly after handling.

Statements: P305, P351 and P338 – IF IN EYES: Rinse with water for several minutes.

Remove contact lenses if present and continue rinsing.

P302 + P352 - IF ON SKIN: Wash with plenty of soap and water.

**HMIS Classification**: Health Hazard 1

Flammability 0 Reactivity 0

Potential HealthInhalation:IrritantEffects:Eye:IrritantSkin:Irritant

Ingestion: Harmful if swallowed

Medical Condition aggravated by long term exposure - Capable of provoking bronchospasm in sulfite sensitive individuals with asthma.



### **SODIUM SULFITE SOLUTION**

# Section 3 - Composition / Information on Ingredients

 Composition
 CAS Number
 % wt or vol

 Sodium Sulfite
 007757-83-7
 15.0 (wt)

 Water
 85.0 (wt)

#### **Section 4 - First Aid Measures**

Exposure Route Symptom Treatment

**Inhalation:** Sore throat, shortness of Remove from exposure to fresh air. Seek

breath coughing, and medical attention in severe cases or if

congestion. recovery is not rapid.

Eye Contact: Irritation to eyes and mucous Irrigate with water until no evidence of

membranes. chemical remains. Obtain medical

attention.

**Skin Contact:** Irritation, itching, dermatitis Wash with soap and drench with water.

Remove contaminated clothing

and wash before reuse.

**Ingestion:** Irritation to mucous membranes. Give large quantities of water or milk

immediately. Obtain medical attention.

After first aid, get appropriate medical attention.

**Note to Physician:** Exposure may aggravate acute or chronic asthma, emphysema and bronchitis.

## **Section 5 - Fire-Fighting Measures**

Flash Point:

Flash Point Method:

Burning Rate:

Not combustible.

Not Applicable.

Not Applicable.

Auto ignition Temperature: Not Applicable.

LEL: Not Applicable.

Vot Applicable.

Not Applicable.

Not Applicable.

Not Applicable.

Not Applicable.

Not Applicable.

Not Applicable.

**Extinguishing Media:** Use extinguishing agent appropriate for surrounding fire conditions.

**Unusual Fire or Explosion** 

**Hazards:** None indicated.

**Hazardous Combustion** 

**Products:** May release hazardous gas.

**Fire-Fighting Instructions:** Do not release runoff from fire control methods to sewers or waterways. **Fire-Fighting Equipment:** Because fire may produce toxic thermal decomposition products, wear a

self-contained breathing apparatus (SCBA) with a full face piece

operated in pressure-demand or positive-pressure mode.



#### SODIUM SULFITE SOLUTION

#### **Section 6 - Accidental Release Measures**

**Spill / Leak Procedures:** Wear appropriate PPE - See Section 8.

Small Spills / Leaks: Leaks may be located by spraying the area with ammonium hydroxide

solution which forms a white fume in the presence of sulfur dioxide.

Large Spills / Leaks: Large spills should be handled according to a predetermined plan.

Containment: For large spills, dike far ahead of contaminated runoff for later

disposal.

Section 7 - Handling and Storage

**Handling Precautions:** Avoid contact with product. Do not breathe vapor.

**Storage Requirements:** Avoid heat or moisture. Store in areas, away from heat and moisture and

protected from physical damage. Segregate from acids and oxidizers.

# Section 8 - Exposure Controls / Personal Protection

Composition	CAS Number	TWA	STEL	IDLH	
Sodium Sulfite	007757-83-7	*	*	*	
Sodium Sulfate	007757-82-6	*	*	*	

<sup>\*</sup> None established. Control as nuisance dust.

**Ventilation:** Provide general or local exhaust ventilation systems to maintain

airborne concentrations below OSHA limits (Sec. 2). Local exhaust ventilation is preferred because it prevents contaminant dispersion

into the work area by controlling it at the source.

**Respiratory Protection:** Follow OSHA respirator regulations (29 CFR 1910.134) and, if necessary,

wear a MSHA/NIOSH-approved respirator. Select respirator based on its suitability to provide adequate worker protection for given working conditions, level of airborne contamination, and presence of sufficient oxygen. For emergency or non-routine operations (cleaning spills, reactor vessels, or storage tanks), wear a SCBA. Warning! Air-purifying respirators do not protect workers in oxygen-deficient atmospheres.

Protective Clothing /

**Equipment:** 

Wear protective gloves, boots, and clothing when necessary to prevent excessive skin contact. Wear protective eyeglasses or goggles, per OSHA eye- and face-protection regulations (29 CFR 1910.133).

Safety Stations: Make emergency eyewash stations, showers, and washing facilities

available in the work area.

**Contaminated Equipment:** 

**Comments:** 

Remove this material from personal protective equipment as needed. Do not eat, drink, or smoke in work areas. Practice good

personal hygiene after using this material, especially before

food or beverage consumption.



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# **Section 9 - Physical and Chemical Properties**

Physical State:LiquidWater Solubility:Completely miscibleAppearance:ClearOther Solubility:Soluble in Glycerin

**Odor Threshold:** Odorless **Boiling Point:** >212 °F <32 <sup>O</sup>F **Vapor Pressure:** NA Freezing Point: Vapor Density (Air=1): **Melting Point:** NA NA Formula Weight: 126.04 **Evaporation Rate:** NA 85 - 95 lb/ft<sup>3</sup> Density: pH: 8 - 9 Specific Gravity (H2O=1): 1.15 % Volatile: NA

# Section 10 - Stability & Reactivity

**Stability:** Stable under normal conditions.

**Polymerization:** Hazardous polymerization will not occur.

Chemical Incompatibilities: Sodium sulfite may, in acidic solutions, release toxic and hazardous

fumes of sulfur oxides, including sulfur dioxide. Acute poisoning from sulfur dioxide is rare because the gas is easily detected. It is so irritating that contact cannot be tolerated. Symptoms include coughing, hoarseness, sneezing, tearing, and breathing difficulty. However, workers who cannot escape high accidental exposure may suffer severe pulmonary damage which can be fatal. Contact with powdered potassium, sodium metals, alkali, and oxidizing agents produce violent reactions. Reacts with water and steam to form corrosive sulfurous acid. Reacts with chlorates to form unstable

chlorine dioxide.

**Conditions to Avoid:** Avoid excessive heat, or open flame.

**Hazardous Decomposition** 

**Products:** May release hazardous sulfur dioxide gas.

#### Section 11 - Toxicological Information

Eye Effects (rabbit): Not available. Acute Inhalation Effects (rabbit): Not Available Skin Effects (rabbit): Not available Acute Oral Effects (rabbit): LD50 = Not Available

**Carcinogenicity:** IARC, NTP, and OSHA do not list Sodium Sulfite as a carcinogen.

**Chronic Effects:** Prolonged or repeated exposure may cause dermatitis, and sensitization reactions.

Exposure to asthmatic, atopic and sulfite sensitive individuals may result in severe bronchoconstriction and reduced levels in forced expiratory volume. Acidic decomposition of sodium sulfite may release toxic and hazardous fumes of sulfur oxides, including sulfur dioxide, which may cause permanent pulmonary impairments

from acute and chronic exposure.



#### **SODIUM SULFITE SOLUTION**

### **Section 12 - Ecological Information**

Ecotoxicity: Sodium Sulfite is non hazardous in solution and is commonly used as a

waste water dechlorination agent. High concentrations will contribute to

elevated chemical oxygen demand in aquatic environments.

**Environmental Transport:** Soluble in water.

Environmental Degradation: Rapid biological decomposition.

Soil Absorption/Mobility: Slight.

### **Section 13 - Disposal Considerations**

Disposal: Waste determinations typically consider Sodium Sulfite contaminated

materials to be non-hazardous.

**Disposal Regulatory** 

**Requirements:** Follow applicable Federal, state and local regulations.

**Container Cleaning and** 

**Disposal:** Follow applicable Federal, state and local regulations.

**Section 14 - Transport Information** 

DOT Transportation Data (49 CFR 172.101):

Shipping Name: Non-Regulated Material

Shipping Symbols: NA
Hazard Class: NA
Subsidiary Hazard: NA
ID No.: NA
Packing Group: NA

Label: GHS label requirements

Special Provisions: None indicated

Section 15 - Regulatory Information

**EPA Regulations:** 

RCRA Hazardous Waste Classification (40 CFR 261): Not listed RCRA Hazardous Waste Number (40 CFR 261): Not listed. CERCLA Hazardous Substance (40 CFR 302.4): Not listed. CERCLA Reportable Quantity (RQ): NA

SARA Title III: Not listed. FIFRA: Not regulated.

TSCA: All components listed

**OSHA Regulations:** 

Air Contaminant (29 CFR 1910.1000)

Not listed.

OSHA Specifically Regulated Substance:

Not listed.

Other Regulations:

FDA (GRAS): Regulated when used as a food preservative.

WHMIS Classification (Canada): D2B

Other Foreign Chemical Control Inventory Listing: Canadian DSL, Australian AICS, Chinese

IECSC, Japanese MITI, Korean KECL, Philippines PICCS and European EINEC.

5



## **SODIUM SULFITE SOLUTION**

### **Section 16 - Other Information**

Previous SDS issue date: May, 2015
Current SDS issue date: September, 2016
Reason for current revision: Company name change

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